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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/522,311	06/30/2005	Kiyotaka Yasuda	8007-1088	5984
466	7590	09/17/2010	EXAMINER	
YOUNG & THOMPSON			HAN, KWANG S	
209 Madison Street				
Suite 500			ART UNIT	PAPER NUMBER
Alexandria, VA 22314			1795	
			NOTIFICATION DATE	DELIVERY MODE
			09/17/2010	ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

DocketingDept@young-thompson.com

Office Action Summary	Application No.	Applicant(s)	
	10/522,311	YASUDA ET AL.	
	Examiner	Art Unit	
	Kwang Han	1795	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

1) Responsive to communication(s) filed on 29 June 2010.
 2a) This action is **FINAL**. 2b) This action is non-final.
 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

4) Claim(s) 13-15 and 17-31 is/are pending in the application.
 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
 5) Claim(s) _____ is/are allowed.
 6) Claim(s) 13-15 and 17-31 is/are rejected.
 7) Claim(s) _____ is/are objected to.
 8) Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

9) The specification is objected to by the Examiner.
 10) The drawing(s) filed on _____ is/are: a) accepted or b) objected to by the Examiner.
 Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
 Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
 a) All b) Some * c) None of:
 1. Certified copies of the priority documents have been received.
 2. Certified copies of the priority documents have been received in Application No. _____.
 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892)	4) <input type="checkbox"/> Interview Summary (PTO-413)
2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948)	Paper No(s)/Mail Date. _____ .
3) <input checked="" type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date <u>6/25/10</u> .	5) <input type="checkbox"/> Notice of Informal Patent Application
	6) <input type="checkbox"/> Other: _____ .

ELECTRODE FOR SECONDARY BATTERY, PROCESS OF PRODUCING THE ELECTRODE, AND SECONDARY BATTERY

Examiner: K. Han SN: 10/522,311 Art Unit: 1795 September 14, 2010

Detailed Action

1. The Applicant's amendment filed on June 29, 2010 was received. Claim 13 was amended. Claims 28-31 were added.
2. The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

Claim Rejections - 35 USC § 112

3. The claim rejections under 35 U.S.C. 112, first paragraph, on claim 13 is withdrawn, because of Applicant's arguments.

Claim Rejections - 35 USC § 102

4. The claim rejections under 35 U.S.C. 102(b) as being anticipated by Kawakami et al. on claims 13, 14, and 17-27 are withdrawn, because the independent claim 13 has been amended.

Claim Rejections - 35 USC § 103

5. Claims 13, 14, and 17-27 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kawakami et al. (US 6051340) in view of Hartman et al. (US 2865973).

Regarding claim 13, Kawakami is directed towards an electrode for a secondary battery [Abstract] comprised of a first and a second surface (Figure 4(a) and 4(d)) both with electrical conductivity and brought into contact with an electrolytic solution (11:38-53), an active material layer containing a powdery material (103) positioned between the first and second surface (10:6-33; Figures 4(c) and Figure 4(d)) subject to a sintering treatment (14:1-3), the powdery material containing alloyable metal (active material) and non-alloyable material (electrically conductive metallic material) mixed together (13:50-66), an electrically metal layer (101) in the middle of the thickness direction, and the active material is present on both sides of the conductive foil (Figure 4(d); 10:25-33) but is silent towards the metallic material to be penetration plated on to the active material particles.

Hartman teaches that an electrode made out of a porous material can be immersed in a plating bath using electrolysis (penetration plating) to deposit through the pores and interstitial surfaces a layer of metal for the benefit of improving the efficiency of the active material and capacity of the battery (1:15-41). It would have been obvious to one of ordinary skill in the art at the time of the invention to use penetration plating to plate the non-alloyable material in the electrode of Kawakami because Hartman teaches the structure formed from this process improves the efficiency of the active material and capacity of the battery.

Regarding claim 14, Kawakami discloses an example of an anode electrode having a total thickness of between 50 to 110 μ m (Columns 21-24).

Regarding claim 17, Kawakami discloses the electrode as an anode [Abstract].

Regarding claims 18 and 19, Kawakami discloses the surface of the anode to be covered by an insulating film (16:23-26) which allows the lithium ions to pass. The layers are formed from a powdery material formed from a sintered body (5:22-30) which would inherently have a porous structure (microvoids) allowing the electrolyte to pass.

Regarding claim 20, Kawakami discloses the formation of the layer comprised of the alloyable (active material particles) and nonalloyable metal with electrically conductive auxiliary to form the anode which has a current collecting function as a whole (13:53-14:36).

Regarding claim 21, Kawakami teaches examples of the electrode having a total thickness as discussed for claim 14. The insulating film is disclosing as a surface layer formed by an insulating film (16:23-26) which has the property of allowing lithium ion to pass but prohibiting lithium metal. This film would inherently have a fractional thickness within the total thickness of the electrode.

Regarding claim 22, Kawakami discloses active material having a metal capable of being alloyed with lithium (10:9-10).

Regarding claims 23 and 24, Kawakami discloses forming the active material layer by obtaining a paste (slurry) and surface layers by electroplating (13:11-14:3).

Regarding claim 25, the teachings of Kawakami as discussed above are herein incorporated. Kawakami further teaches a nonalloyable metal (material with “lower” capability of forming a lithium compound) in the active material layer (13:39-46).

Regarding claim 26, Kawakami discloses a multilayer structure of the anode layer where the metal incapable of being alloyed with lithium is at an enhanced content

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at the anodes surface forming a multilayer surface in combination with the insulating film (5:13-21; Figure 4d).

Regarding claim 27, Kawakami discloses the powdery material comprising the active material layer are mixed together (13:50-66) which would inherently have some degree of porosity (vacant spaces) due to the nature of the powder material.

Regarding claims 28 and 30, Kawakami discloses examples where the active material particles have a mean particle size of 300 mesh (approximately 49 microns).

5. Claim 15 is rejected under 35 U.S.C. 103(a) as being unpatentable over Kawakami et al. and Hartman et al. as applied to claim 13 above and further in view of Beard (US 5147739).

Regarding claim 15, the teachings of Kawakami and Hartman as discussed above are herein incorporated. Kawakami discloses active material having a metal capable of being alloyed with lithium (103) (10:9-10) but is silent towards a metallic lithium layer provided between the conductive foil and the active material layer.

Beard teaches a lithium battery comprised of having an anode with a current collector 13, metallic lithium layer 14, and an active material layer 15 (Figure 1A) for the benefit of providing a electrochemical cell with the full voltage available from a pure lithium anode without the problems of dendritic growth or lithium cycling loss (5:25-6:6). It would have been obvious to one of ordinary skill in the art at the time of the invention have a metallic lithium layer between the conductive foil and active material layer of Kawakami because Beard teaches that this configuration provides a battery which has

the full voltage available from a pure lithium electrode without the problems of dendritic growth or lithium cycling loss.

6. Claims 29 and 31 are rejected under 35 U.S.C. 103(a) as being unpatentable over Kawakami et al. and Hartman et al. as applied to claims 13 and 20 above, and further in view of Gosho et al. (US 6589694).

The teachings of Kawakami as discussed above are herein incorporated.

Regarding claims 29 and 31, Kawakami discloses the conductive material to be acetylene black (14:22-25) but is silent towards the particles size to be 0.1 microns and the active material particles to have to have a size less than 20 microns.

Gosho teaches the active material particles for a lithium secondary battery should be between 10-25 microns to avoid abnormal cell reactions and minimize electric resistance (6:31-38) and further teaches conductive material such as acetylene black should have a particle size 2 microns or smaller so that the particles gather mainly on the surface of the active material to improve the conductivity (8:25-27; 9:21-28). It has been held that where the claimed ranges “overlap or lie inside ranges disclosed by the prior art” a prima facie case of obviousness exists. In re Wertheim, 541 F.2d 257, 191 USPQ 90 (CCPA 1976); In re Woodruff, 919 F.2d 1575, 16 USPQ2d 1934 (Fed. Cir. 1990) (MPEP 2144.05). It would have been obvious to one of ordinary skill in the art at the time of the invention to provide active material particles between 10-25 microns and the conductive particles sized 2 microns or smaller because Gosho teaches this allows the battery to avoid abnormal cell reactions, minimize electric resistance, and improve conductivity.

Response to Arguments

7. Applicant's arguments with respect to claims 13-15 and 17-31 have been considered but are moot in view of the new ground(s) of rejection.

Conclusion

8. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Contact/Correspondence Information

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Kwang Han whose telephone number is (571) 270-5264. The examiner can normally be reached on Monday through Friday 8:00am to 5:00pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Dah-Wei Yuan can be reached on (571) 272-1295. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/K. H./
Examiner, Art Unit 1795

/Dah-Wei D. Yuan/
Supervisory Patent Examiner, Art Unit 1795